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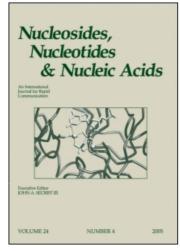
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## Nucleosides, Nucleotides and Nucleic Acids

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# Aryl Nucleoside H-Phosphonates—Novel Derivatives of Controlled Reactivity

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# ARYL NUCLEOSIDE H-PHOSPHONATES - NOVEL DERIVATIVES OF CONTROLLED REACTIVITY

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**ABSTRACT**: The most essential factors influencing the formation of aryl nucleoside H-phosphonates are discussed.

Recently, aryl nucleoside H-phosphonates (1) (Scheme 1) emerged as a new type of intermediates in the synthesis of phosphate derivatives<sup>1</sup>. In contradistinction to other reactive species derived from H-phosphonate monoesters, these compounds bear only one electrophilic centre (located on phosphorus) and their reactivity can be modulated by substituents on the aromatic ring<sup>2</sup>.

In this report we wish to discuss factors influencing the formation of aryl nucleoside H-phosphonates under various experimental conditions. Reactivity of phenols in a coupling reaction with H-phosphonate monoesters 2 varied, and sometimes provided different products depending on a pKa value of the phenol used. In a standard condensation promoted by pivaloyl chloride in pyridine phenols with pKa values higher than 9.0 (e.g. 4-chlorophenol, phenol and 4-methylphenol)<sup>3</sup> produced exclusively the expected aryl nucleoside H-phosphonate 1. Phenols with 9.0 > pKa > 6.0 (e.g. 2,4-dichloro-, 4-nitro-, 2,4,6-trichloro) afforded under these conditions variable amounts of tervalent bis-aryl nucleoside phosphites of type 3, along with the desired 1. Pentachlorophenol (pKa < 6), however, produced in a coupling reaction with 2 exclusively 3.

938 CIEŚLAK ET AL.

### SCHEME1

 $\mathbf{R}$  = suitably protected nucleoside;  $\mathbf{Ar}$  = aryl;  $\mathbf{X}$  = pivaloyl or diphenylphosphate residue

The coupling reaction with phenols of moderate acidity (pKa > 6) performed in less basic media, e.g. CH<sub>2</sub>Cl<sub>2</sub>/pyridine 9 : 1 (v/v) produced exclusively aryl nucleoside H-phosphonates 1. This is most likely due to suppression of a double activation of 2 (to produce 4<sup>4</sup>) under these conditions and thus, predominantly monoactivated species 5 are formed. Besides pivaloyl chloride, the most convenient condensing agent for this purpose was found to be diphenyl phosphorochloridate. With this reagent (1.5 molar equiv.) aryl nucleoside H-phosphonates 1 were formed fast (ca. 3 min) and quantitatively.

Very acidic phenols ( pKa < 6, e.g. pentachloro-, pentafluorophenol and 2,4-dinitrophenol) in  $CH_2Cl_2$ /pyridine 9:1 (v/v) also produced aryl nucleoside H-phosphonates of type 1. These, however, were not completely stable under the reaction conditions and underwent a gradual disproportionation towards bis-aryl nucleoside phosphite of type 3 and the initial nucleoside H-phosphonate 2, according to the mechanism we have described earlier<sup>5</sup>.

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